

BOARD QUESTION PAPER: MARCH 2020

Science and Technology Part - 1

Time: 2 Hours

Total Marks: 40

- Note:**
- i. All questions are compulsory.
 - ii. Use of a calculator is not allowed.
 - iii. The numbers to the right of the questions indicate full marks.
 - iv. In case of MCQs (Q. No. 1(A)) only the first attempt will be evaluated and will be given credit.
 - v. For each MCQ, the correct alternative (A), (B), (C), (D) with sub-question number is to be written as an answer.
For Eg: (i) (A), (ii) (B), (iii) (C)
 - vi. Scientifically correct, labelled diagrams should be drawn wherever necessary.

Q.1. (A) Write the correct alternative: [5]

- i. According to Mendeleev's periodic law, properties of elements are periodic function of their _____.
(A) Atomic numbers (B) Atomic masses
(C) Densities (D) Boiling points
- ii. The vapour content in the air is measured using a physical quantity called _____.
(A) Absolute humidity (B) Relative humidity
(C) Dew point (D) Humidity
- iii. For the normal human eye, the near point is at _____ cm.
(A) 10 (B) 20 (C) 25 (D) 30
- iv. The astronomical object closest to us is _____ is our galaxy.
(A) Mars (B) Venus (C) Jupiter (D) Moon
- v. In the Wilfley table method, the particles of gangue are separated by _____ separation method.
(A) Magnetic (B) Froth floatation (C) Hydraulic (D) Gravitational

(B) Answer the following: [5]

- i. Find the odd one out:
Voltmeter, Ammeter, Thermometer, Galvanometer.
- ii. Complete the correlation:
Alkene : C = C :: Alkyne: _____.
- iii. State true or false:
The frequency of AC is 50 Hz.
- iv. Match the Columns:

Column 'A'	Column 'B'	
The wavelength of red light	(a)	600 nm
	(b)	700 nm
	(c)	500 nm

- v. Name the first artificial satellite sent by Russia in space.

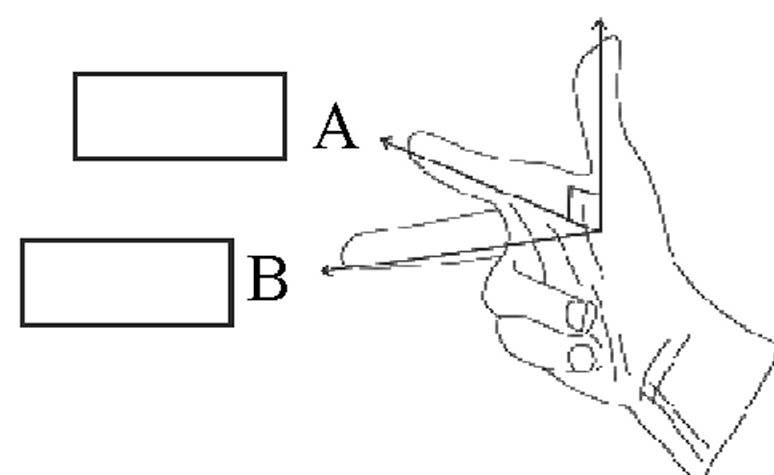
Q.2. (A) Give scientific reasons (any two): [4]

- i. The weight of an object changes from place to place though its mass is constant.
- ii. Stars twinkle but we do not see the twinkling of planets.
- iii. Elements belonging to the same group have the same valency.

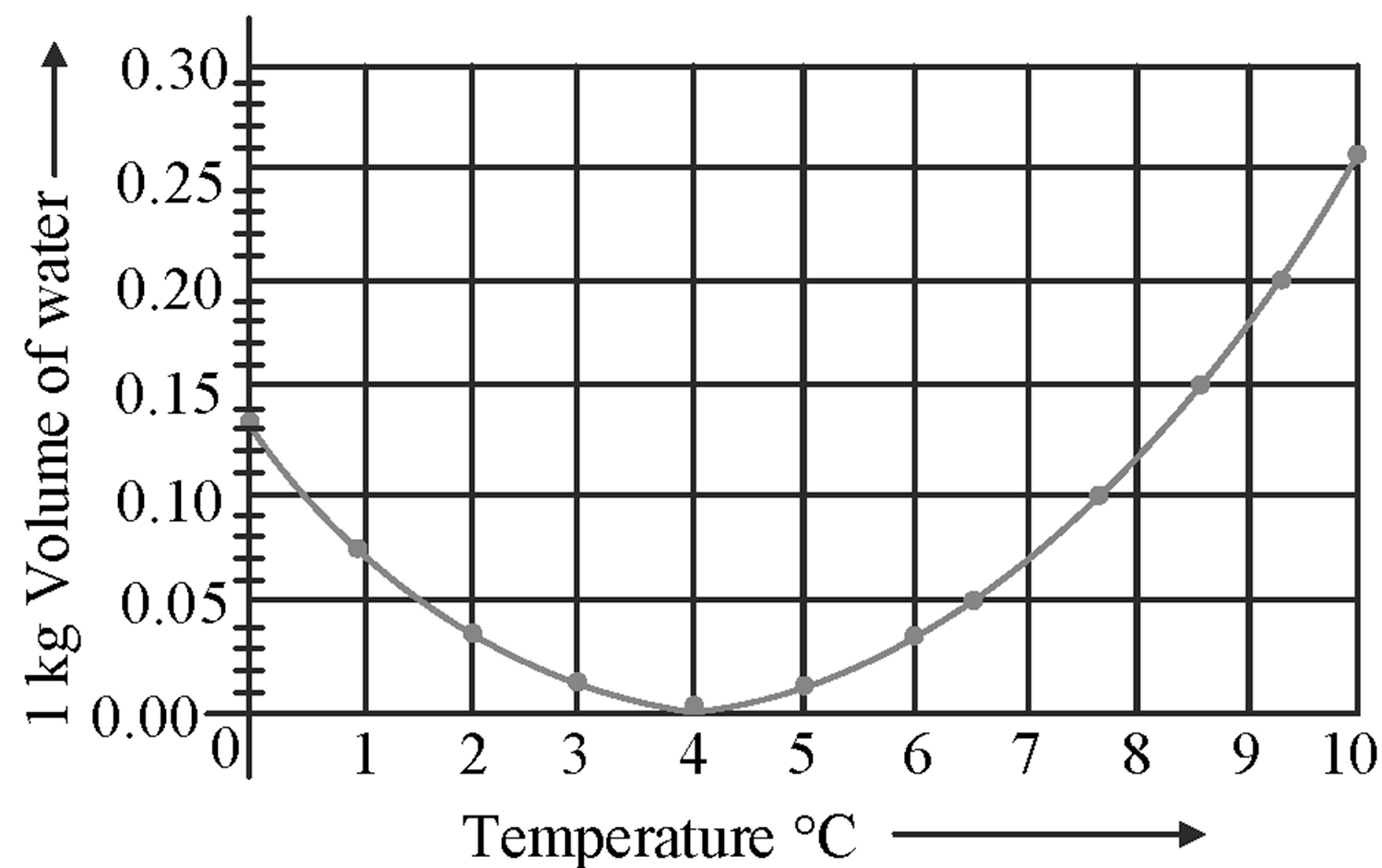
(B) Answer the following (any three): [6]

- i. How much heat energy is necessary to raise the temperature of 5 kg of water from 20°C to 100°C?
- ii. Observe the given figure of Fleming's Right Hand Rule and write the labels of A and B correctly.

Motion of a conductor

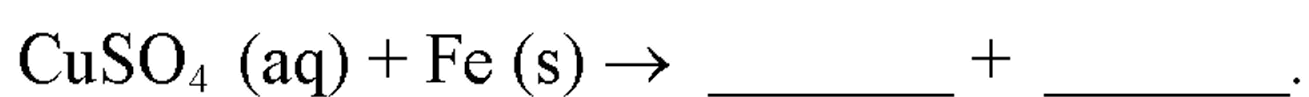


iii. Observe the given graph and answer the following questions:



- Name the process represented in the figure.
- At what temperature does this process take place?

iv. Complete the given chemical reaction:



Name the type of the reaction.

v. Write a short note on Alloying.

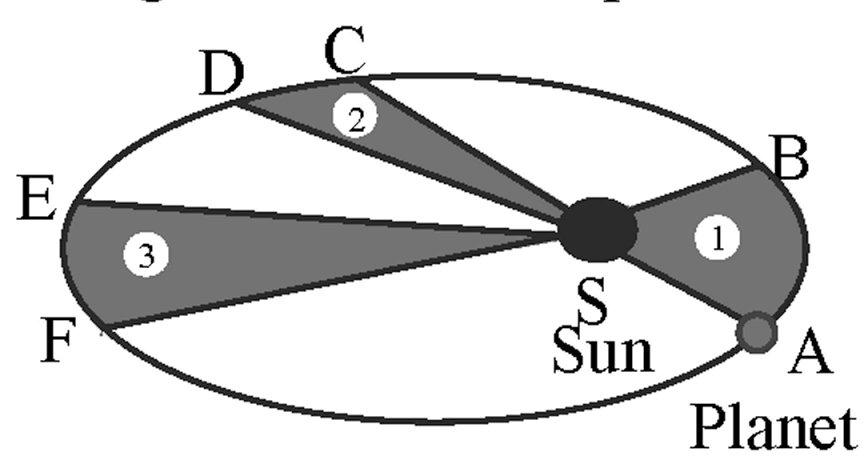
Q.3. Answer the following (any five):

[15]

i. An element has its electronic configuration as 2, 8, 2. Now answer the following questions:

- What is the atomic number of this element?
- What is the group of this element?
- To which period does this element belong?

ii. Observe the given figure showing the orbit of a planet moving around the Sun and write the *three* laws related to it:



The orbit of a planet moving around the Sun

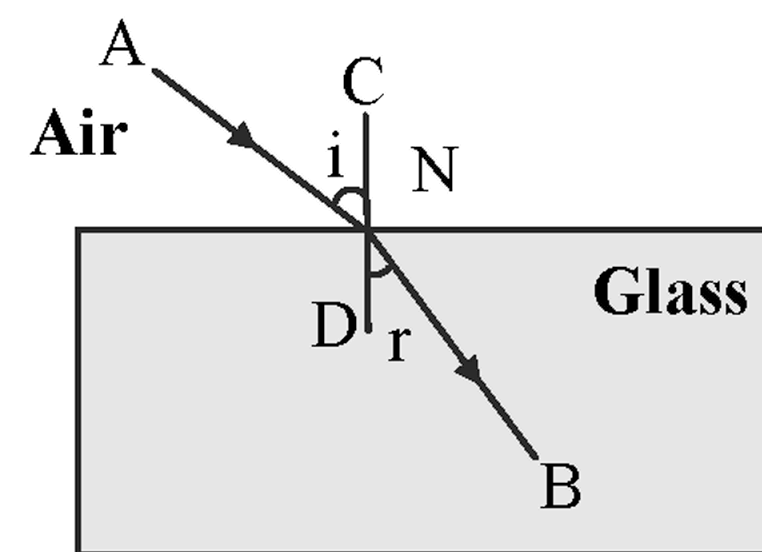
iii. Read the given passage and answer the following questions:

The home electrical connection consists of 'live', 'neutral' and 'earth' wires. The 'live' and the 'neutral' wires have potential difference of 220 V. The 'earth' is connected to ground. Due to a fault in the equipment or if the plastic coating on the 'live' and the 'neutral' wires gives a way the two wires come in contact with each other and a large current flows through it producing heat. If any inflammable material (such as wood, cloth, plastic, etc.) exists around that place it can catch fire. Therefore a fuse wire is used as a precautionary measure.

- Name the two wires having potential difference of 220 V.
- What is short circuit?
- Write the function of a fuse.

iv. Observe the given figure and answer the following questions:

- Name the process represented by the figure.
- State the two laws related to the process.

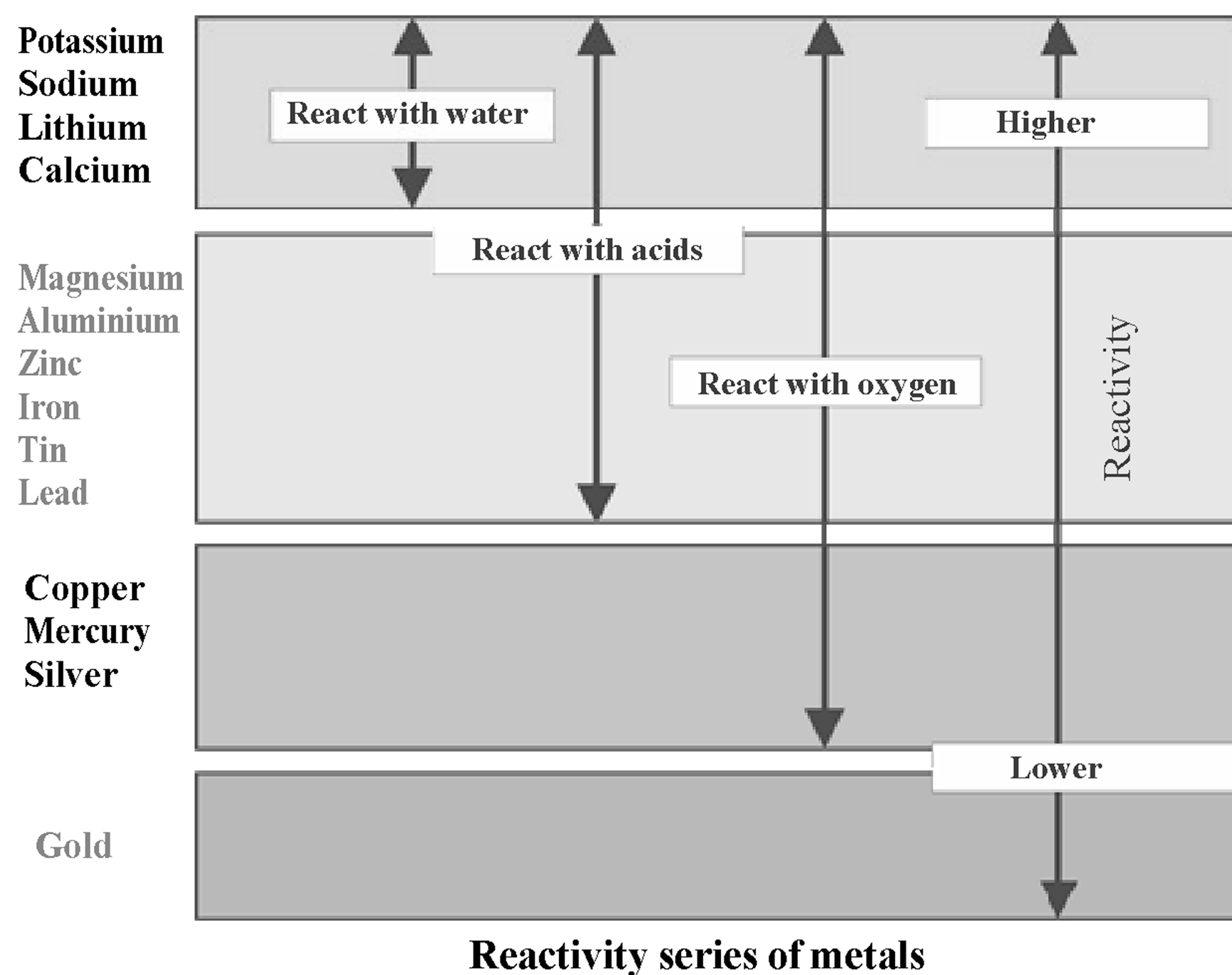


v. What is an artificial satellite? Name any two types of artificial satellite and state their functions.

vi. Answer the following questions:

- Define Hydrocarbons.
- Name the types of Hydrocarbons.
- Name two carbon compounds used in day-to-day life.

vii. Observe the given figure of reactivity series of metals and answer the following questions:



- Name two metals which react with water.
- Name two moderately reactive metals.
- Name the most highly reactive metal and the most less reactive metal.

viii. Complete the following table:

Straight chain of Carbon compounds	Structural formula	Molecular formula	Name
C	$\begin{array}{c} \text{H} \\ \\ \text{H} - \text{C} - \text{H} \\ \\ \text{H} \end{array}$	CH ₄	Methane
C-C	_____	_____	Ethane
C-C-C	_____	C ₃ H ₈	_____
C-C-C-C	$\begin{array}{cccc} \text{H} & \text{H} & \text{H} & \text{H} \\ & & & \\ \text{H} - \text{C} & - \text{C} & - \text{C} & - \text{C} - \text{H} \\ & & & \\ \text{H} & \text{H} & \text{H} & \text{H} \end{array}$	_____	_____

Q.4. Answer any one of the following:

[5]

- Draw a scientifically correct labelled diagram of a human eye and answer the questions based on it:
 - Name the type of lens in the human eye.
 - Name the screen at which the maximum amount of incident light is refracted?
 - State the nature of the image formed of the object on the screen inside the eye.

ii. Observe the following picture and answer the following questions:

- What is a rust?
- Write the chemical formula of rust.
- Write the reaction of oxidation of iron at anode.
- Write the reaction of oxidation of iron at cathode.
- What is corrosion?

